

Thermal Physics

Energy of a particle in a 1D box

Consider a box of one dimension.

The kinetic energy is: $E = \frac{\hbar^2}{2mL^2} n^2$, where $n=1, 2, 3, \dots$

The average energy is: $\langle E \rangle = \frac{\sum_{n=1}^{\infty} \frac{\hbar^2}{2mL^2} n^2 \exp\left(-\frac{\hbar^2}{2mL^2 k_B T} n^2\right)}{\sum_{n=1}^{\infty} \exp\left(-\frac{\hbar^2}{2mL^2 k_B T} n^2\right)}$

Approximating the summations with integrals:

$$\langle E \rangle = k_B T \frac{\sum_{n=1}^{\infty} \frac{\hbar^2}{2mL^2 k_B T} n^2 \exp\left(-\frac{\hbar^2}{2mL^2 k_B T} n^2\right)}{\sum_{n=1}^{\infty} \exp\left(-\frac{\hbar^2}{2mL^2 k_B T} n^2\right)} \approx k_B T \frac{\int_0^{\infty} u^2 \exp(-u^2) du}{\int_0^{\infty} \exp(-u^2) du}$$

Both, the numerator and denominator are Gaussian integrals with values:

$$\int_0^{\infty} \exp(-u^2) du = \frac{\sqrt{\pi}}{2} \quad \text{and} \quad \int_0^{\infty} u^2 \exp(-u^2) du = \frac{\sqrt{\pi}}{4} \quad \text{then:}$$

$$U = \langle E \rangle \approx \frac{1}{2} k_B T$$