

# Quantum Mechanics

## Bosons and Fermions

**Problem 1.-** Classify the following neutral atoms as fermions or bosons:



**Solution:** We need to count the number of fermions (electrons, protons and neutrons) if the number is even, the neutral atom is a boson, otherwise a fermion:



**Problem 2.-** Write down a symmetric wave function of three identical bosons, using the wave functions  $\psi_1$ ,  $\psi_2$  and  $\psi_3$ .

**Solution:** If we sum all the possible combinations of the three wave functions we will get a symmetric wave function with parity +1:

$$\begin{aligned} &\psi_1(\vec{r}_1)\psi_2(\vec{r}_2)\psi_3(\vec{r}_3) + \psi_1(\vec{r}_1)\psi_3(\vec{r}_2)\psi_2(\vec{r}_3) + \psi_2(\vec{r}_1)\psi_1(\vec{r}_2)\psi_3(\vec{r}_3) + \psi_2(\vec{r}_1)\psi_3(\vec{r}_2)\psi_1(\vec{r}_3) + \\ &\psi_3(\vec{r}_1)\psi_1(\vec{r}_2)\psi_2(\vec{r}_3) + \psi_3(\vec{r}_1)\psi_2(\vec{r}_2)\psi_1(\vec{r}_3) \end{aligned}$$

To normalize the wave function, we need to divide it by  $\sqrt{6}$ .

**Problem 3.-** Write down an antisymmetric wave function of three identical fermions, using the wave functions  $\psi_1$ ,  $\psi_2$  and  $\psi_3$ .

**Solution:** We could generate a Slater determinant and write down an antisymmetric wave function:

$$\begin{vmatrix} \psi_1(\vec{r}_1) & \psi_1(\vec{r}_2) & \psi_1(\vec{r}_3) \\ \psi_2(\vec{r}_1) & \psi_2(\vec{r}_2) & \psi_2(\vec{r}_3) \\ \psi_3(\vec{r}_1) & \psi_3(\vec{r}_2) & \psi_3(\vec{r}_3) \end{vmatrix}$$

To normalize the wave function, we need to divide it by  $\sqrt{6}$ .

**Problem 4.-** A system containing two identical particles is described by a wave function of the form:

$$\psi = \frac{1}{\sqrt{2}} [\psi_a(x_1)\psi_b(x_2) - \psi_a(x_2)\psi_b(x_1)]$$

Where  $x_1$  and  $x_2$  represent the spatial coordinates of the particles and  $a$  and  $b$  represent all the quantum numbers, including spin, of the states that they occupy. The particles might be:

- (A) Deuterons.
- (B) Neutral sodium atoms  ${}^{23}_{11}\text{Na}$
- (C) Alpha particles
- (D) Neutral nitrogen atoms  ${}^{14}_7\text{N}$
- (E) Neutral rubidium-85 isotopes  ${}^{85}_{37}\text{Rb}$

**Solution:** The wave function given has parity -1, that is, the wave function changes sign when two particles are swapped:

$$\psi = \frac{1}{\sqrt{2}} [\psi_a(x_1)\psi_b(x_2) - \psi_a(x_2)\psi_b(x_1)]$$

This means that it represents fermions.

The only fermions in the list are neutral nitrogen atoms  ${}^{14}_7\text{N}$ .

**Answer: D**